## ON THE PROPERTIES OF SOME SUBSTITUTED $\alpha$ '-LITHIATED

 $\alpha(Z)$ , Y-BUTADIENYL SULFIDES: CYCLISATIONS AND REARRANGEMENTS

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Summary. The conditions for the preparation and cyclisation of the title compounds  $\underline{3}$  are described. The stability of the intermediate lithio-thiacyclohexenes  $\underline{4}$  is shown to depend on the reaction conditions and the substituents  $R^1-R^4$ . The final products  $\underline{6}$ ,  $\underline{10}$  and  $\underline{7}$  arise from  $\underline{4}$  via a [1.4] sigmatropic rearrangement, a [1.3] prototropic shift followed by a [2.3] sigmatropic rearrangement, or the elimination of lithium methane-thiolate respectively.

We recently reported the stereoselective conversion of lithiated benzylic 3-methyl-1(Z),3-butadienyl sulfide  $\underline{3}a$  (R<sup>2</sup>=H) into the cis-disubstituted cyclopropane  $\underline{5}a$  <sup>1</sup>. In order to verify that this product  $\underline{5}a$  arises from the lithiated thiacyclohexene  $\underline{4}a$  (R<sup>2</sup>=H) by a [1.4] sigmatropic rearrangement <sup>2</sup>, we studied the metalation of thiacyclohexene  $\underline{2}a$  (R<sup>2</sup>=R<sup>5</sup>=H). Entry 1 shows that under certain conditions, the carbanion  $\underline{4}a$  (R<sup>2</sup>=H) can be trapped with methyl iodide. When this carbanion  $\underline{4}a$  (R<sup>2</sup>=H) was submitted to the same conditions as for  $\underline{3}a$  (R<sup>2</sup>=H)(HMPA/-30°C/CH<sub>3</sub>I), the same cyclopropane compound  $\underline{6}$  was obtained as major product but with a small amount of another cyclopropane compound  $\underline{10}a$  (entry 6).

When the thiacyclohexene 2a ( $R^2=R^5=H$ ) was present in excess relative to the base, the new cyclopropane compound 10a 3 was obtained as the sole product with an equal amount of the starting material (entry 11). Likewise, when the sulfide 1a ( $R^2=H$ ) was present in excess relative to the base, the same cyclopropane compound 10a was obtained with an equal amount of the thiacyclohexene 1a (1a0 (entry 10). These results can be explained by intermolecular proton shifts 4. As soon as it is formed, the thiacyclohexene 1a1 (1a2 (1a3 soon as it is formed, the thiacyclohexene 1a3 soon as it is formed, the thiacyclohexene 1a4 (1a3 soon as it is formed, the thiacyclohexene 1a4 soon as it is formed, the thiacyclohexene 1a4 so 1a5 soon as it is formed, the thiacyclohexene 1a6 so 1a6 so 1a7 so 1a8 soon as it is formed, the thiacyclohexene 1a8 so 1a9 s

- by the starting material  $\underline{1}a$  ( $R^2$ =H) to give the benzylic carbanion  $\underline{3}a$  ( $R^2$ =H) which cyclises to  $\underline{4}a$  ( $R^2$ =H) and the cycle repeats;
- by the thiacyclohexene 2a (R<sup>2</sup>=R<sup>5</sup>=H) when the dienic sulfide is absent or consumed; this new thiacyclohexene lithio-derivative 8a can then undergo a classical [2.3] sigmatropic rearrangement to give 9a, the precursor of the cyclopropane compound 10a.

Table

Entry	Substrate	Conditions	yield %*	Products (proportions)
1	<u>2</u> a R <sup>2</sup> =R <sup>5</sup> =H	ii; quenching at -78°C with CH <sub>3</sub> I	90	$2^{\alpha}$ (R <sup>2</sup> =H, R <sup>5</sup> =CH <sub>3</sub> ) (t/c = 75/25)
2	<u>1</u> b R <sup>2</sup> =H	i; idem	80	$1b (R^2 = CH_3)$
3	<u>l</u> c R <sup>2</sup> =H	i; warming to -30°C then 2 h at -30°C and quenching with CH <sub>3</sub> I	(99)	$\underline{1}_{c} (R^2 = CH_3)$
	<u>1</u> b R <sup>2</sup> =H	i; warming to -30°C then 2 h at -30°C and quenching with H <sub>2</sub> 0	60	$\underline{2}$ b (R <sup>5</sup> =H) (t/c = 75/25)
5	<u>1</u> ь R <sup>2</sup> =Н	idem, quenching with CH3I	62	$\underline{2}$ b (R <sup>5</sup> =CH <sub>3</sub> ) (three isomers 64/18/18)
6	<u>2</u> a R <sup>2</sup> =R <sup>5</sup> =H	ii; 1.2 equiv HMPA added, warming to -30°C then 2 h at -30°C and quenching with CH <sub>3</sub> I	90	<u>6</u> (78); <u>10</u> a (22)
7	<u>l</u> b R <sup>2</sup> =H	i; idem	51	$\underline{2}$ b ( $R^2$ =H, $R^5$ =CH <sub>3</sub> ) (15); $\underline{10}$ b (85)
8	<u>1</u> c R <sup>2</sup> =H	iii; idem	71	<u>10</u> c
9	$1c R^2 = H$	iv; quenching with H <sub>2</sub> 0	46	<u>7</u>
10	<u>l</u> a R <sup>2</sup> =H	v; 1.2 equiv HMPA added, warming to -30°C then 2 h at -30°C and quenching with CH <sub>3</sub> I	(93)	$ \underline{2}a (R^2 = R^5 = H) (50); $ $ \underline{10}a (50) $
11	$2a R^2 = R^5 = H$	vi; idem	(95)	$\underline{2}a (R^2 = R^5 = H) (50);$ $\underline{10}a (50)$
12	<u>l</u> c R <sup>2</sup> =H	v; idem	(98)	$\underline{2}c (R^2 = R^5 = H) (50);$ $\underline{10}c (50)$

<sup>\*</sup> The numbers in brackets are the yields of crude products.

i 1.2 equiv of  $\text{n.C}_4\text{H}_9\text{Li}$  was added to a cooled (-78°C) solution of the substrate in THF and the mixture stirred at that temperature for 30 min.

ii same as i, but with  $sec.C_4H_9Li$  instead of  $n.C_4H_9Li$ .

iii same as i, but with lithium diisopropylamide (LDA) instead of n. C<sub>A</sub>H<sub>Q</sub>Li.

iv same as i followed by addition of 1.2 equiv of  $n.C_4H_9I$  and warming to  $18^{\circ}C$ ; then at -78°C, 1.2 equiv of  $n.C_4H_9Li$  added, 30 min; then 1.2 equiv HMPA added, warming to -30°C and 2 h at -30°C.

v same as i, but 0.5 equiv of  $n.C_4H_9Li$  instead of 1.2 equiv.

vi same as ii, but with 0.5 equiv of  $\sec.C_4H_9Li$  instead of 1.2 equiv.

Series	$R^1$	$R^2$	$R^3$	$R^4$	<sub>₹</sub> 5
a	<sup>С</sup> 6 <sup>Н</sup> 5	H or CH3	Н	СН3	H or CH <sub>3</sub>
b	<sup>С</sup> 6 <sup>Н</sup> 5	н	-(CH	1 <sub>2</sub> ) <sub>4</sub> -	H or CH <sub>3</sub>
С	CH <sub>3</sub> -S	H or CH <sub>3</sub> or n.C <sub>4</sub> H <sub>9</sub>	Н	CH <sub>3</sub>	-

Two new cyclopropane compounds  $\underline{10}$ b,c were obtained by cyclisation and rearrangement of (Z)-dienyl sulfides  $\underline{1}$ b (R<sup>2</sup>=H) and  $\underline{1}$ c (R<sup>2</sup>=H) respectively. These (Z)-dienyl sulfides were prepared in yields of 70 and 48 % by the nucleophilic addition of thiol to the enyne under basic conditions 5.

At -30°C, the lithiated sulfide 3b ( $R^2$ =H) underwent smooth cyclisation to the stable lithiated thiacyclohexene 4b (trans/cis = 75/25)(entries 4, 5); however, when 3b was treated with 1.2 equiv of HMPA at the same temperature and quenched with methyl iodide, it did not give the product of a [1.4] sigmatropic rearrangement, but afforded the cyclopropane compound 10b as the major product (entry 7).

Finally, the isoprenic dithioacetal  $\underline{lc}$  ( $R^2$ =H) afforded three different products depending on the reaction conditions:

- when treated with a slight excess of LDA and HMPA, then with methyl iodide, it was smoothly converted into the cyclopropanone dithioacetal 10c (entry 8) c;
- when 1c ( $R^2$ =H) was present in excess relative to the base, it yielded analogous to 1c ( $R^2$ =H), a mixture of the thia-cyclohexene 1c ( $R^2$ =R) and the cyclopropane 1c (entry 1c);
- and when treated with a slight excess of n.butyllithium followed by n.butyl iodide and then again with n.butyllithium and HMPA, it gave a lithio-thiacyclohexene  $\underline{4}$ c ( $R^2$ =n. $C_4$ Hg) which, unable to undergo the aforementioned proton shift leading to vinyl-cyclopropane  $\underline{10}$ , suffers loss of lithium methanethiolate  $\overline{7}$  to afford  $\overline{7}$  as the sole product (entry 9).

A full account of this work including additional examples and a detailed mechanistic analysis will be published elsewhere.

## REFERENCES AND NOTES

- 1. M. Reglier, S.A. Julia, Tetrahedron Letters, 1983, 24, 2387.
- 2. J.F. Biellmann, J.B. Ducep, J.J. Vicens, <u>Tetrahedron</u>, 1976, <u>32</u>, 1801.
- 3. All new compounds exhibited the expected  $\overline{IR}$ , N.M.R. ( $^{1}H$ ,  $^{13}C$ ) and mass spectral properties.
- 4. As it was reported for the lithiation of allyl benzyl sulfide: J.F. Biellmann, J.B. Ducep, D. Schirlin, Tetrahedron, 1980, 36, 1249.
- R.H. Everhardus, R. Gräfing, L. Brandsma, <u>Synthesis</u>, 1983, 623; M. Reglier, O. Ruel,
   R. Lorne, S.A. Julia, <u>ibidem</u>, 1983, 624.
- 6. The compound <u>10</u>c is analogous to those obtained by [2.3] sigmatropic rearrangement of appropriate sulfonium ylides: E.J. Corey, S.W. Walinsky, <u>J. Amer. Chem. Soc.</u>, 1972, 94, 8932.
- 7. A similar ring closure to a cyclopropane compound due to an intramolecular displacement of phenylthiolate ion by an allylic aliphatic lithio-derivative has been reported: O. Ruel, B. Cazes, S. Julia, <u>Synthetic Cammun.</u>, 1980, <u>10</u>, 743.

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